

# Multiscale Model-Based Process Systems Engineering

## Progress and Challenges

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## Outline

- An overview of multiscale modeling
- Drivers of nanomaterials research/New frontiers
- Control in self-assembly
- Design of materials for energy and sustainability
- Outlook



## Outline

- **An overview of multiscale modeling**
- Drivers of nanomaterials research/New frontiers
- Control in self-assembly
- Design of materials for energy and sustainability
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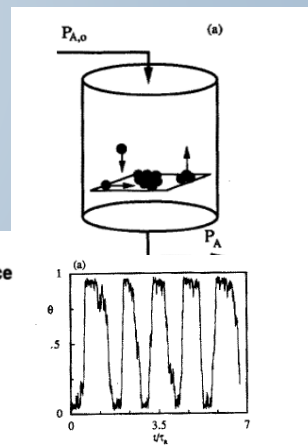
## Some Esoteric Problems in Nonlinear Dynamics

- The first demonstration of hybrid multiscale modeling
  - ❖ **Spatial KMC coupled with ODEs**

### The effects of phase transitions, surface diffusion, and defects on surface catalyzed reactions: Fluctuations and oscillations

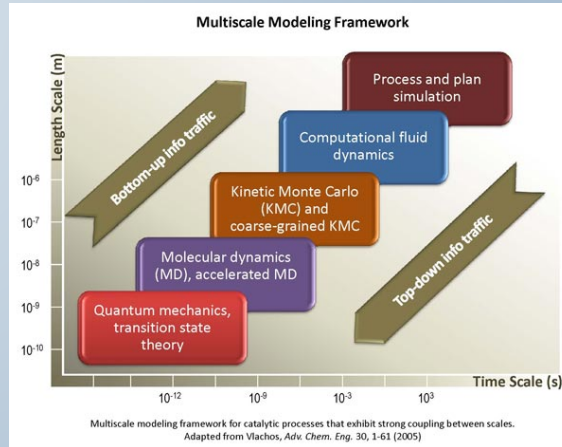
D. G. Vlachos, L. D. Schmidt,<sup>90</sup> and R. Aris

8306 J. Chem. Phys. 93 (11), 1 December 1990 0021-9606/90/238306-08\$03.00 © 1990 American Institute of Physics

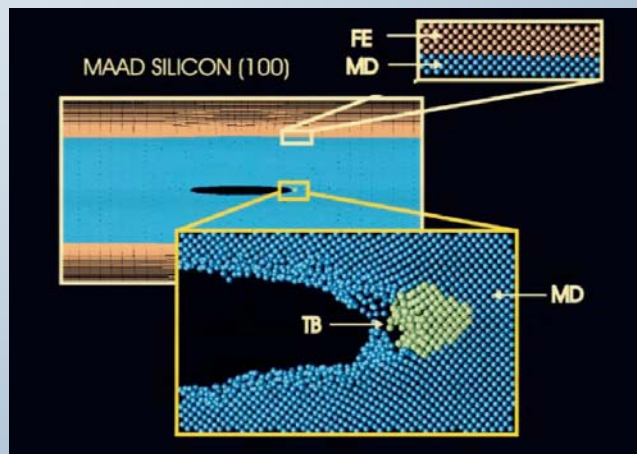


# Multiscale Modeling

- Mathematical & computational methods developed
  - ❖ Bottom up modeling
    - **Process engineering**



# Materials Mechanics (Dislocation, Crack Initiation)



See also quasi-continuum method: Tadmor et al. (1996), Phillips (1998)

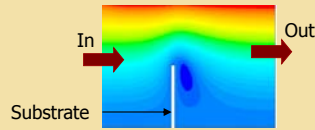
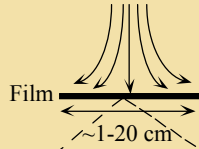
Broughton et al., PRB 1999

# Fluid – Surface Coupling

## Gas-phase continuum model/Finite difference

Continuum heat & mass transfer, fluid mechanics equations for a stagnation flow geometry

Expt parameters:  
Composition, flow rate, P, T

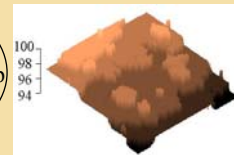
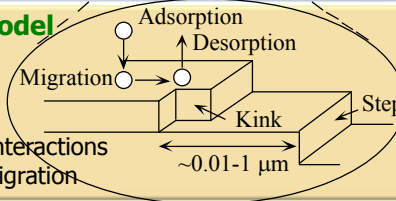


**Dynamic coupling through fluxes**

## Surface atomistic model

### KMC simulation

Solid-on-solid lattice  
First-nearest neighbors interactions  
Adsorption-Desorption-Migration

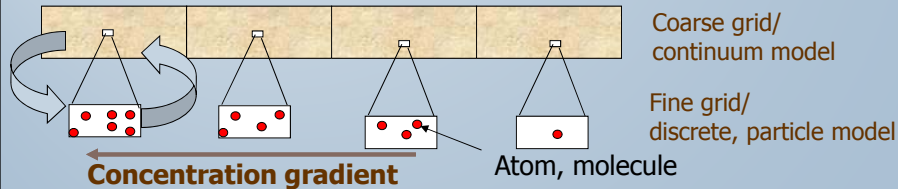


See also subsequent work on control of thin film roughness by Christofides and Cu electrodeposition by Braatz

Vlachos, *Appl. Phys. Lett.* 74, 2797 (1999)

# Bridging the Gap of Length and Time Scales

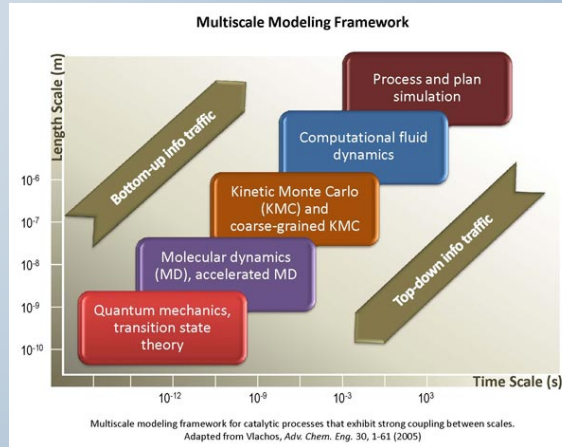
- The **tooth-gap** (Kevrekidis) and **heterogeneous multiscale method** (E) were introduced
  - ❖ Different models are used at different grids
  - ❖ Pass dynamically info between grids
  - ❖ **Speedup due to length and time scale disparity**
    - **Simulate small lengths for short periods with the micro solver**



Gear et al., *Phys. Letters A* 316, 190 (2003); E et al., *Phys. Rev. B* 67, 092101 (2003)

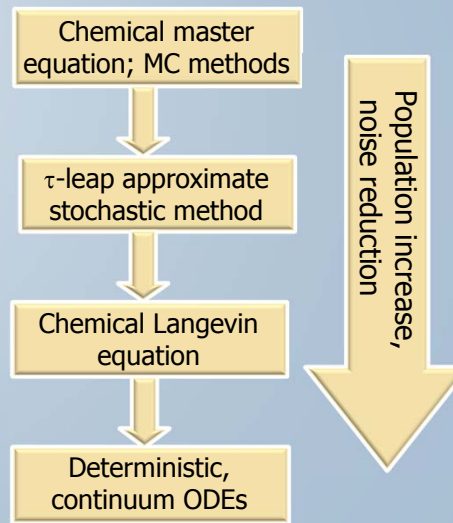
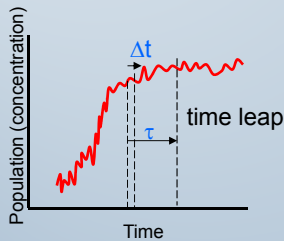
# Multiscale Modeling

- Mathematical & computational methods developed
  - ❖ Bottom up modeling
    - **Process engineering**
    - ❖ **Coarse-graining**



# Temporally Coarse-Grained KMC

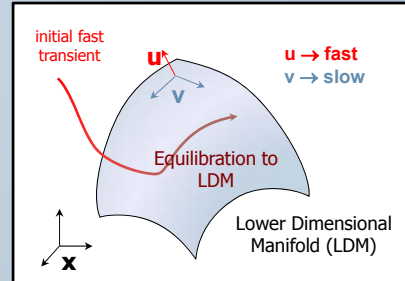
- Monte Carlo (MC) methods (Gillespie, 1975; 1976)
- **A hierarchy of models exists for well-mixed systems**



Review: Chatterjee and Vlachos, *J. Comp.-Aided Mater. Des.* **14**, 253 (2007)

## Temporally Coarse-Grained KMC

- Monte Carlo (MC) methods (Gillespie, 1975; 1976)
- A hierarchy of models exists for well-mixed systems
- **Stochastic low dimensional manifold (SLDM): overcome stiffness<sup>1</sup>**
  - ❖ CSP assisted partitioning<sup>2</sup>
  - ❖ Convergence to LDM<sup>2</sup>



<sup>1</sup> Gillespie, Petzold, E, Kaznessis, Vlachos, etc.

<sup>2</sup> Samant and Vlachos, *J. Chem. Phys.* **123**, 144114 (2005)

Review: Chatterjee and Vlachos, *J. Comp.-Aided Mater. Des.* **14**, 253 (2007)

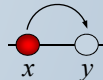
## A Hierarchy of Coarse Grained KMC Simulations

### Microscopic Stochastic Process

$$\text{Hamiltonian: } H(\sigma) = -\frac{1}{2} \sum_{x \in M} \sum_{y \neq x} J(x-y) \sigma(x) \sigma(y) + \sum_{x \in M} h \sigma(x)$$

Microscopic processes/transition

$$v(x \rightarrow y, \sigma) = v_0 \sigma(x) (1 - \sigma(y)) \exp[-\beta(U_0 + U(x))]$$



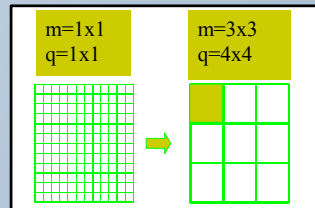
### Mesoscopic Stochastic Process

$$\eta_k = \sum_{y \in D_k} \sigma(y) \quad q = \text{number of points in a coarse cell}$$

$$\bar{U}_i = \sum_{k \in M_c} \bar{J}_{ik} \eta_k + \bar{J}_{00} (\eta_i - 1) - h$$

$$\bar{H}(\eta) = - \sum_{k \in M_c} \bar{U}_k \eta_k = -\frac{1}{2} \sum_{i \in M_c} \sum_{\substack{k \in M_c \\ k \neq i}} \bar{J}_{ik} \eta_k \eta_i - \frac{\bar{J}_{00}}{2} \sum_{i \in M_c} \eta_i (\eta_i - 1) + \sum_{k \in M_c} h \eta_k$$

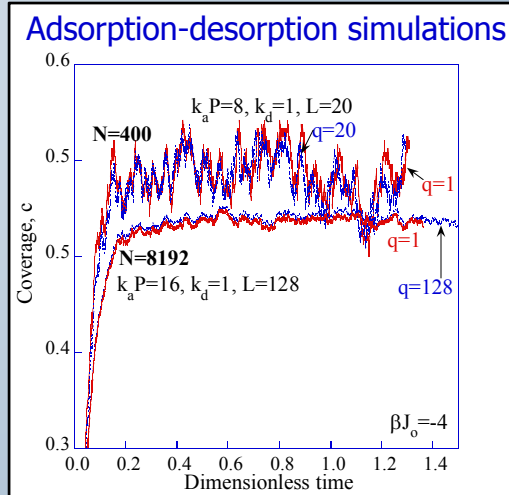
$$\bar{v}(k \rightarrow i, \eta) = v_0 \eta_k (q - \eta_i) \exp[-\beta(U_0 + \bar{U}_k)]$$



*J. Comp. Phys.* **186**, 250 (2003)

## The Only CG Method with Correct Noise

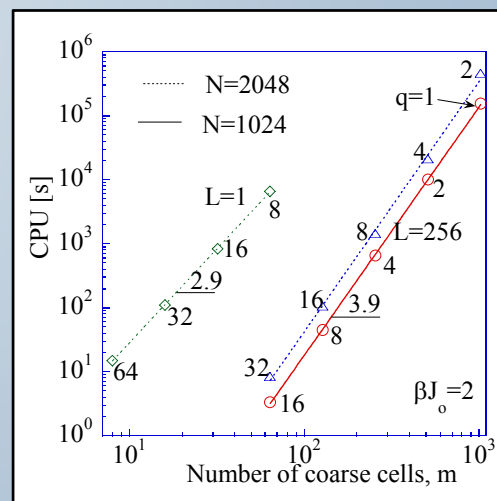
- The noise is correctly simulated by coarse-grained KMC simulations at the same length scale
- The Gibbs states and the Large Deviation Principles of the coarse-grained and the microscopic processes are asymptotically (long potential) identical



Katsoulakis et al., *PNAS* **100**, 782 (2003)

## Profound CPU Savings Enable Simulation of Realistic Scales

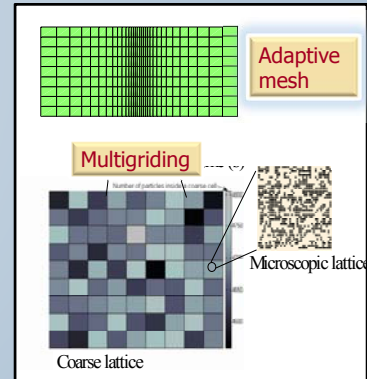
- $\text{CPU} \sim 1/q^{2-4}$ 
  - ❖ 4-5 sites per dimension in 3D, leads to  $10^8$  reduction of CPU!
  - ❖ Overcome hydrodynamic slowdown (diffusion):  $q^2$
  - ❖ More savings can result by employing local update search algorithms for coarse potentials
- Simulation of large length scales is feasible



Katsoulakis and Vlachos, *J. Chem. Phys.* **119**, 9412 (2003)

## Overview of Spatial Acceleration Methods

- Spatial adaptivity<sup>1</sup>
  - ❖ A posteriori error estimates guide mesh refinement and generation of phase diagrams
- High accuracy
  - ❖ Higher order closures<sup>2</sup>
  - ❖ Multigrid/Relaxation criteria<sup>2</sup>
  - ❖ Rigorous cluster expansions/Theory<sup>3,4</sup>
- Multicomponent systems<sup>5,6</sup>
- Time scale acceleration



<sup>1</sup> Chatterjee et al., *JCP* **121**, 11420 (2004); *PRE* **71**, 0267021 (2005); Katsoulakis, Plechac, et al., *J. Non Newtonian Fluid Mech.* (2007)

<sup>2</sup> Chatterjee and Vlachos, *JCP* **124**, 0641101 (2006)

<sup>3</sup> Katsoulakis, Plechac, et al., *ESAIM, Math Model. Num. Anal.* **41**, 627-660 (2007)

<sup>4</sup> Katsoulakis, Plechac, Sopasakis, *SIAM Num. Anal.* (2006)

<sup>5</sup> Chatterjee and Vlachos, *JCP* **127**, 034705 (2007)

<sup>6</sup> Collins, Chatterjee and Vlachos, *JCP*, **129**, No. 184101 (2008)

Review: Chatterjee and Vlachos, *J. Comp.-Aided Mater. Des.* **14**, 253 (2007)

## Continuum Mesoscopic Equations

- Mesoscopic equation (for diffusion)

$$\partial c_t = -\nabla \cdot (-\mu[c] \nabla (\delta\phi[c] / \delta c))$$

$$\frac{\partial c}{\partial t} = \nabla \cdot \{ D e^{-\beta J^* c} [\nabla c - \beta c(1-c) \nabla J^* c] \}$$

$$J^* c = \iint J(|r - r'|) c(r') dr'$$

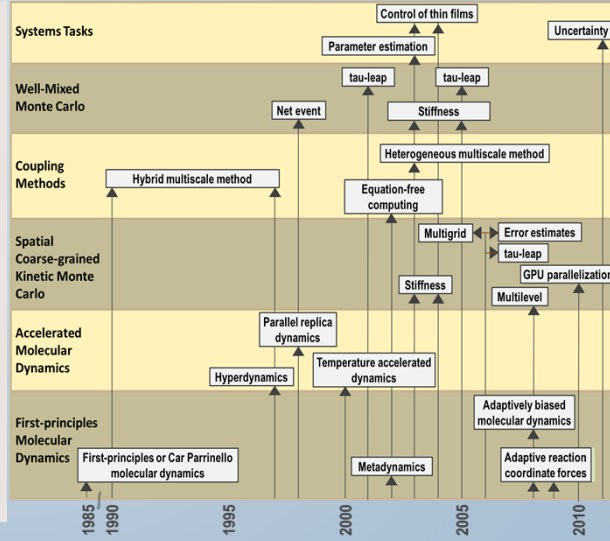
- Mesoscopic equations include essential physics
  - ❖ Intermolecular forces, microscopic mechanisms
  - ❖ Are exact when interactions are long ranged
- Amenable to systems tasks
- Spectral methods for efficient, parallelizable calculations

*PRL* **85**, 3898 (2000); *PRL* **84**, 1511-1514 (2000); *Chem. Eng. Sci.* **58**, 895 (2003)



# The 30,000 Miles Airview

- Significant progress made on method development and testing
- Focus has been on **prototype problems**
- Field is maturing
- **Much less work has been done at the systems' level**



Vlachos, *AIChE J.* **58(5)**, 1314 (2012)

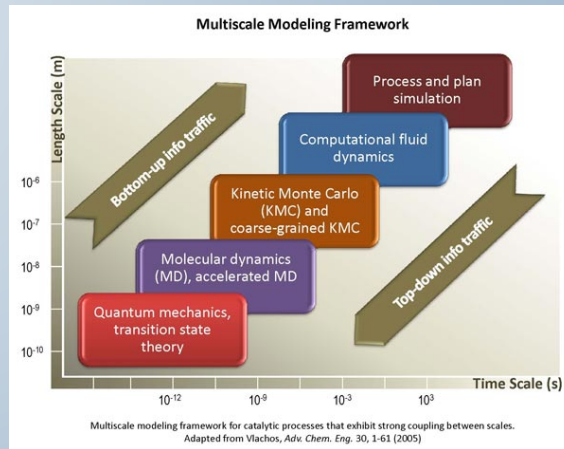
# Outline

- An overview of multiscale modeling
- **Drivers of nanomaterials research/New frontiers**
- Control in self-assembly
- Design of materials for energy and sustainability
- Outlook



## Multiscale Modeling

- Mathematical & computational methods developed
  - ❖ Bottom up modeling
    - **Process engineering**
  - ❖ Coarse-graining
  - ❖ **Top down modeling**
    - **Product engineering**



## APOLLO 13: Houston We Have a Problem

- Multiscale modeling methods from prototype systems do not apply/extend well to real problems
- Real problems are hard!

## Realistic Nanomaterials Process Modeling: New Frontiers

- Product design and control at the nanoscopic scale
- Emergent behavior and/or emergent properties
- Combinatorial complexity
- Validation, verification, and uncertainty
- Computer architectures

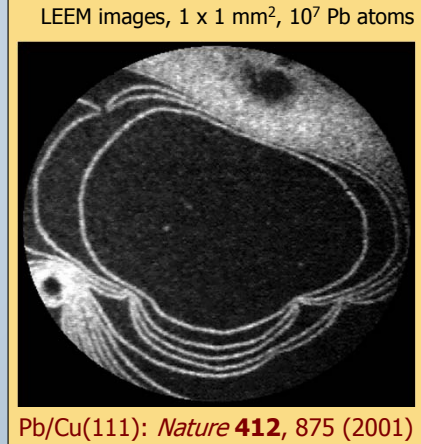
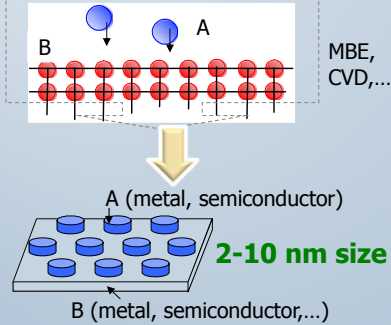
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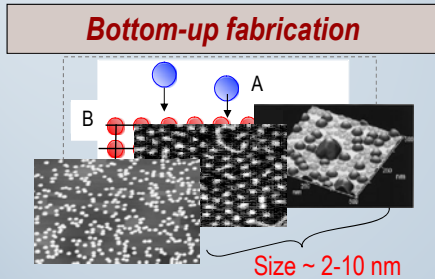
## Self-Assembly in Heteroepitaxy

- Self-organization is a bottom up strategy for materials synthesis
  - ❖ Low cost; broad applications (optical, magnetics, electronics, catalysts,...)

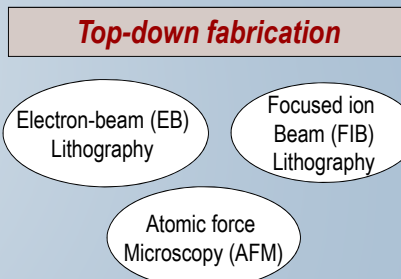


- **Non-uniform shape, size, spacing obtained**
- **How does one 'control' structures and thereby enable scaling up?**

## Bottom - up Fabrication is Attractive



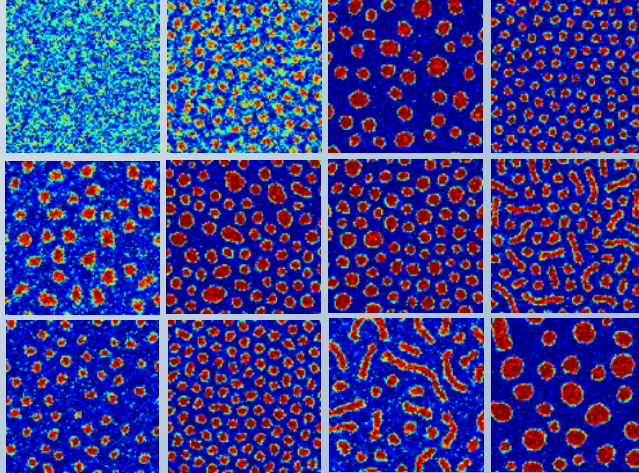
- Stable minimum free energy structures
- **Small feature size**
- Use of existing deposition technologies
- **Scalable, economically viable**
- **Disordered structures**



- Electron-beam (EB) Lithography
- Focused ion Beam (FIB) Lithography
- Atomic force Microscopy (AFM)
- Strain defects often formed
- EB, FIB: **Feature size > 45 nm**
- Need for new deposition technologies
- AFM: **Not scalable**, not economically viable

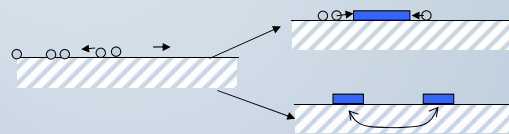
## Emergent Behavior

- Collective behavior not predicted from atom-atom interactions
- Patterns very sensitive to operating conditions



## Role of Interactions in Self-Assembly

- Patterns form by the interplay of short attractive and **long repulsive interactions**

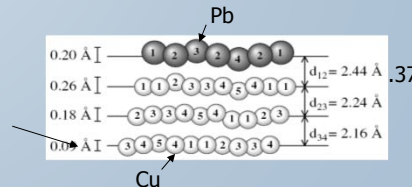


Elastic repulsion > 20 lattice constants long

$$J(r) = j_0 \left( e^{-(r/r_s)^2} - \chi e^{-(r/r_r)^2} \right)$$

$$\chi \approx \frac{Ea^3}{1-\nu^2} \left( \frac{\Delta a}{a} \right)^2$$

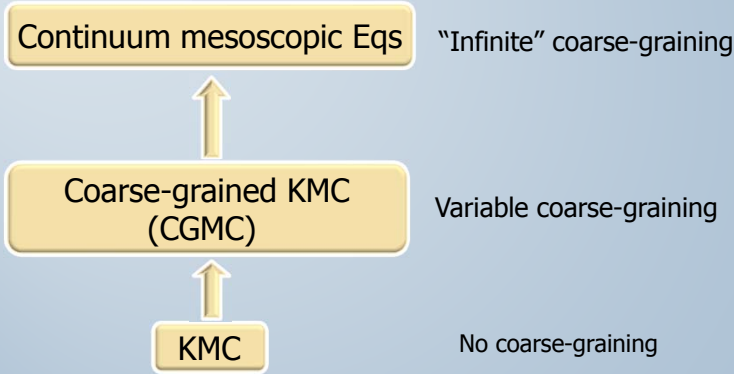
- Repulsive interactions
  - ❖ Lattice-mismatch
  - ❖ Electronic effects



J. Phys. Condensed Matter 13 (2001) 1793

- **Computational cost prohibitive for atomistic KMC**

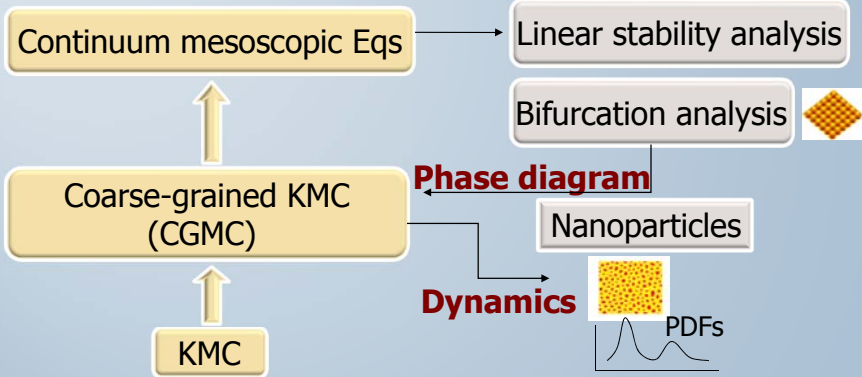
## Hierarchical Multiscale Modeling for Pattern Formation



**Bottom-up tool development**

PRL **85** (2000) 3898; JCP **119** (2003) 9412; PNAS **100**, 782 (2003); JCP **124** (2006) 64110

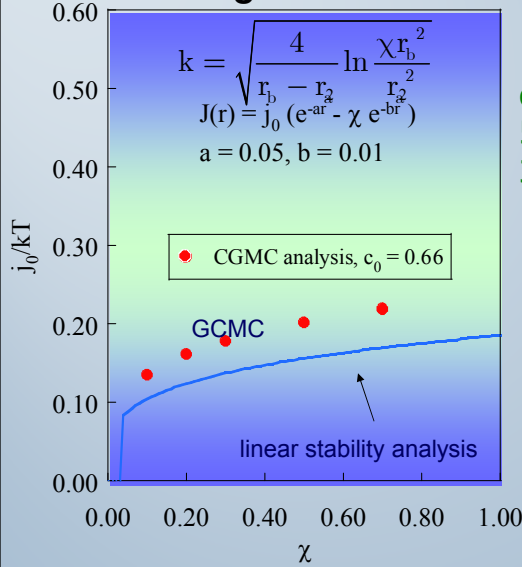
## Hierarchical Multiscale Modeling for Pattern Formation



**Top-down screening**

Chatterjee and Vlachos, *CES* **62(18-20)**, 4852 (2007)

## Phase Diagram: Linear Stability vs. GCMC



**q**    **CPU**    **CPU reduction**  
 5x5    11 hr    0.4x10<sup>6</sup> times  
 3x3    3-4 d    7x10<sup>3</sup> times

$\lambda = 12 - 22 \text{ nm}$

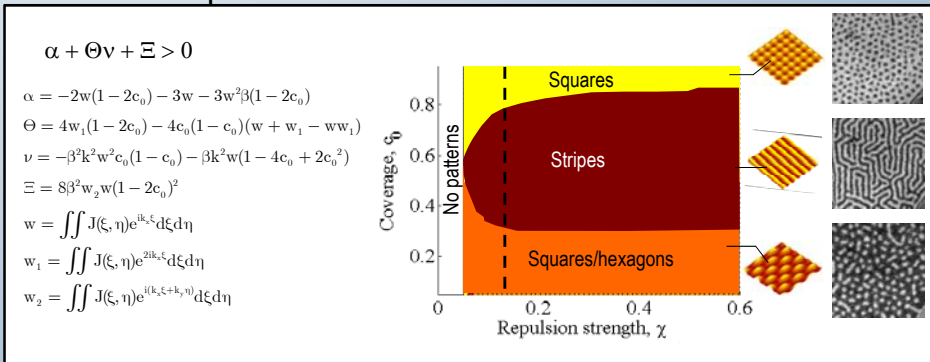
$N = 300 \times 300$  lattice sites

$a = 1 \text{ nm} \Rightarrow 0.3 \mu\text{m} \times 0.3 \mu\text{m}$

Chatterjee and Vlachos, *CES* **62(18-20)**, 4852 (2007)

## Patterns' Shape: Nonlinear Analysis

Different shapes form under certain conditions

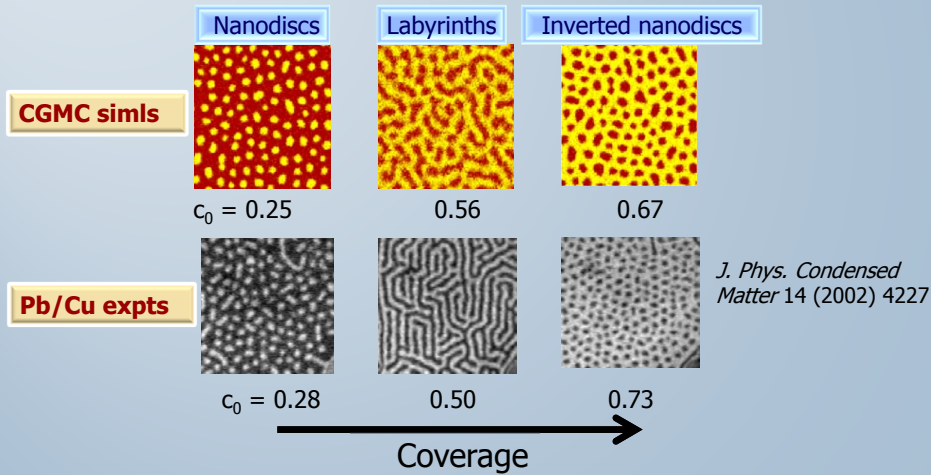


Exp: *J. Phys. Condensed Matter* 14 (2002) 4227

- Patterns are in qualitative agreement with experimental data
- Key differences are: (1) labyrinths, (2) order-disorder

Chatterjee and Vlachos, *CES* **62(18-20)**, 4852 (2007)

## CGMC Simulations Mimic Experiments



Chatterjee and Vlachos, *CES* **62(18-20)**, 4852 (2007)

## Open Questions

- Hierarchy of length and time scales
  - ❖ Process model-based control impossible
- Emergent behavior
  - ❖ Sluggish dynamics, sensitivity to conditions
- Extremely high dimension (distributed parameter systems)
- Intrinsic stochastic fluctuations
- Poor controllability
  - ❖ Control nanoscale features over several orders of magnitude
  - ❖ A mere handful of manipulated variables at the macroscale
- Poor observability
  - ❖ Measurements of features are unavailable online

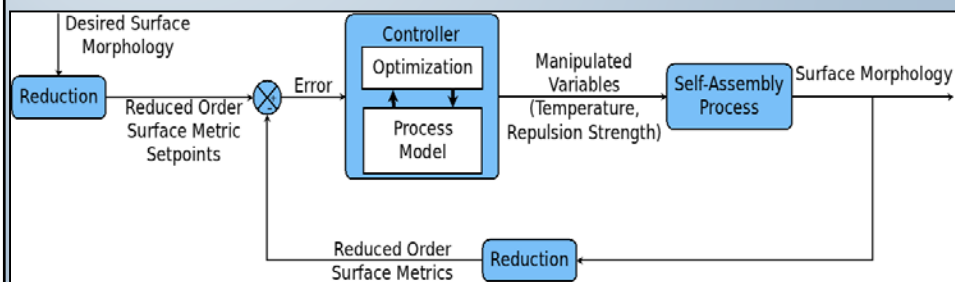


## Systems Tasks Needed

- Overall control strategy
- Gradient estimation
- Sensitivity analysis
- Efficient and robust optimization algorithms (agent based)
- Quantitative metrics to compare with experiments

## Overarching Approach

- Determine the degree of controllability
- Implement a control strategy
- Account for statistical information in experiments and models (pdfs)



## Finite Differencing Gradient Estimation

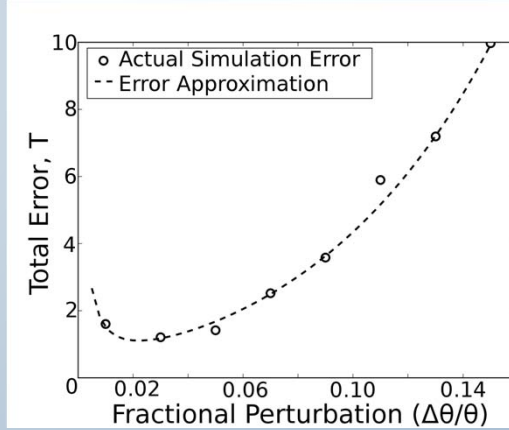
- Easy to implement
- FD gradient estimate

$$E(G) = \frac{E(f(\theta + \Delta\theta)) - E(f(\theta - \Delta\theta))}{2\Delta\theta} + O(\Delta\theta^2)$$

$$\text{var}(G) = \frac{1}{(2\Delta\theta)^2} [\text{var}(f|_{\theta+\Delta\theta}) + \text{var}(f|_{\theta-\Delta\theta})]$$

- Total error:

$$T = \left( \frac{\text{var}(G)}{n} \right)^{1/2} + O(\Delta\theta^2)$$



- **There is an optimal choice for  $\Delta\theta$**
- CPU: Linear scaling with number of parameters
- Variance reduction methods: CNR and CRP

McGill et al., *J. Comput. Phys.* **231**, 7170(2012)

## Likelihood Ratio Gradient Estimation

- A weighted average similar to importance sampling

$$\frac{\partial E(f)}{\partial \theta_j} = E \left( f \frac{\partial \ln p_v}{\partial \theta_j} \right)$$

$$\frac{\partial \ln p_v}{\partial \theta_j} = \sum_{i=1}^n \left[ \frac{I(\mu_i, j)}{\theta_j} - \tau_i h_j(S_i) \right]$$

$$I(\mu_i, j) = \begin{cases} 1 & \text{if } \mu_i = j \\ 0 & \text{if } \mu_i \neq j \end{cases}$$

$I(\mu_i, j)$  : indicator function  
 $\mu_i$  : reaction fired at  $i^{\text{th}}$  event  
 $\tau_i$  : time step of  $i^{\text{th}}$  event  
 $S_i$  : state at  $i^{\text{th}}$  event  
 $h_j(\cdot)$  : related to propensity of  $j^{\text{th}}$  reaction  
 $p_v$  : joint pdf for random variables

- **Calculate the entire gradient vector from one simulation**
- Trivial code modification (record reaction fired and time)

McGill et al., *J. Comput. Phys.* **231**, 7170(2012)

## LRGE vs. FD Estimation Methods

- Likelihood ratio method is unbiased compared to FD methods
- LRGE is a better estimator despite having a large variance
- LRGE requires fewer simulation evaluations

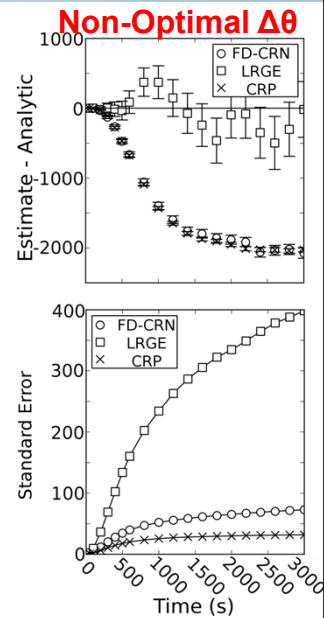
LRGE: Likelihood ratio gradient estimation<sup>[1]</sup>

FD-CRN: FD with common random numbers

CRP: Common reaction path method<sup>[2]</sup>, low variance FD

[1] J.A. McGill et al., J. Comput. Phys. (2012)

[2] M. Rathinam et al., J. Chem. Phys. (2010)



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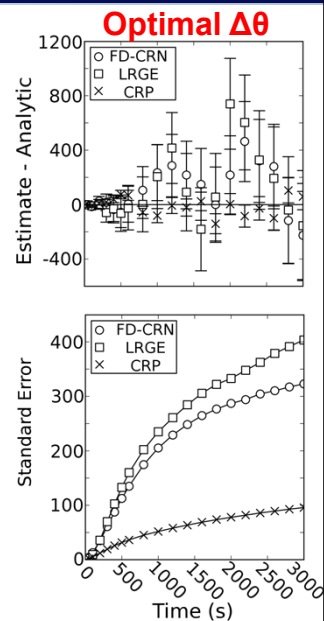
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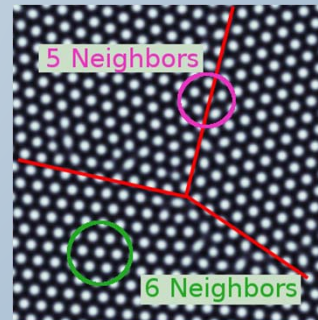


## Define Metrics

- Difficult to use images as process outputs for control
  - ❖ Translations and rotations: *morphologically equivalent* but *different images*
- Morphological information arises from **image analysis**
- Define **metrics** that are *invariant to basic transforms* on the image

### Metrics (9)

- ✓ Hexagonal order parameter
- ✓ Distance between islands
- ✓ Characteristic length
  - ✓ From spectral analysis of surface<sup>[1]</sup>
- ✓ Number of defects
- ✓ Minkowski measures<sup>[2]</sup>
  - ✓ Area fraction
  - ✓ Perimeter of nanodots
  - ✓ Euler characteristic (connectivity)

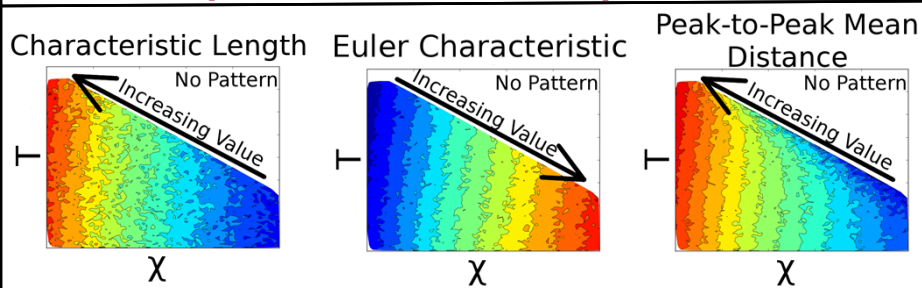


Abukhdeir and Vlachos, J. Comp. Phys. (2011); [2] Legland et al., Image Anal Sterol (2007)

## Eliminate Correlated Metrics

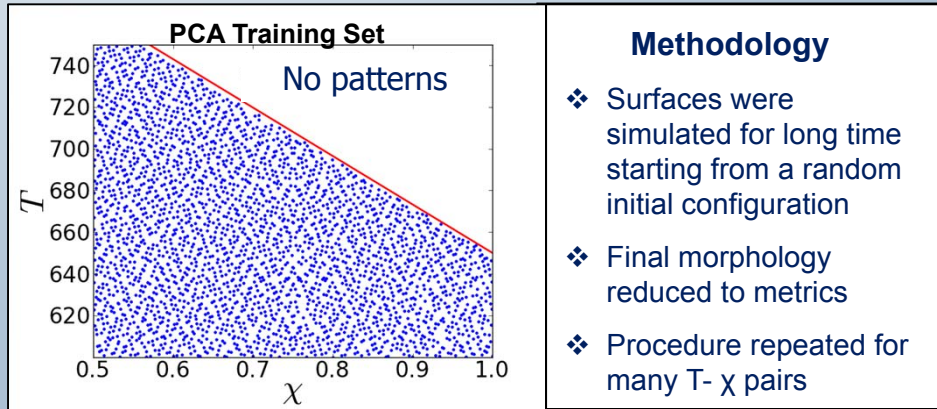
- Many surface metrics (9) exist but are **not** necessarily **independent**
- **Cannot use metrics directly** as process outputs
  - ❖ Correlations cause *instability* in controller

### Contour plots of metrics vs. manipulated variables



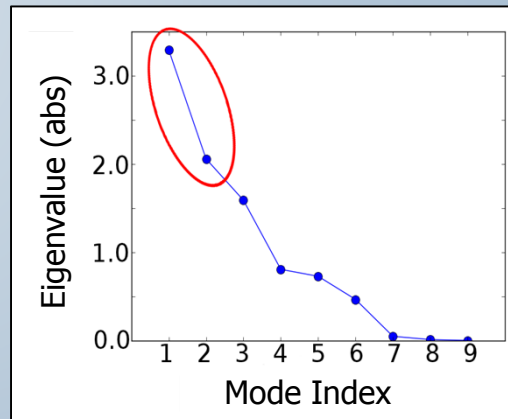
## Image Reduction via PCA

- Use Principal Component Analysis (PCA) to find an **orthogonal basis** for surface metrics



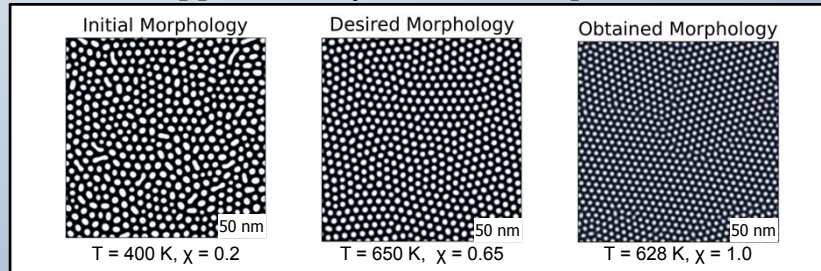
## Image Reduction via PCA

- Find an orthogonal basis for surface metrics
- Use PCA modes as process outputs
  - ❖ Orthogonality implies *some controllability*
  - ❖ Unaccounted modes treated as a disturbance
- First two modes dominate



## PCA Modes in a Simple PI Controller

- PCA modes can be used as process observables
- Are they *good* observables?
  - ❖ Proof of concept with a simple PI controller
  - ❖ Assumed approximately first order response



- **Control is possible** (error reduced by order of magnitude)
  - ❖ Refinement of controller should yield better results

## Open Questions

- Hierarchy of length and time scales
  - ❖ Process model-based control impossible
- Emergent behavior
  - ❖ Sluggish dynamics, sensitivity to conditions
- Extremely high dimension (distributed parameter systems)
- Intrinsic stochastic fluctuations
- Poor controllability
  - ❖ Control nanoscale features over several orders of magnitude
  - ❖ A mere handful of manipulated variables at the macroscale
- Poor observability
  - ❖ Measurements of features are unavailable online

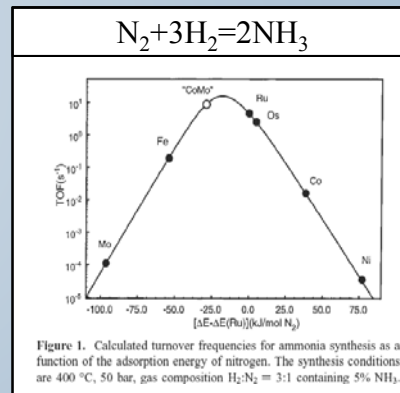
## Outline

- An overview of multiscale modeling
- Drivers of nanomaterials research/New frontiers
- Control in self-assembly
- **Design of catalytic materials for energy and sustainability**
- Outlook



## Predicting Novel Catalytic Materials

- First principles methods (DFT) promise to deliver strategies for rational catalyst design
- Current studies limited to cases when:
  - Thermodynamics (heat of adsorption) dominates<sup>1,2</sup>
  - Linear interpolation is employed<sup>3</sup>

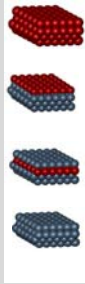
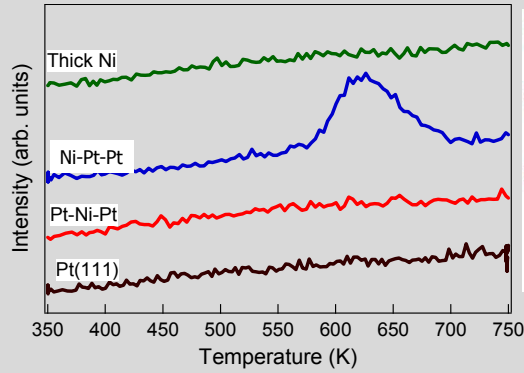


<sup>1</sup>Strasser et al., *J. Phys. Chem. B* **107**(40), 11013 (2003)

<sup>2</sup>Greeley and Mavrikakis, *Nat. Materials* **3**(11), 810 (2004)

<sup>3</sup>Jacobsen et al., *J. Am. Chem. Soc.* **123**, 8404 (2001)

## Molecular Architecture Plays a Pivotal Role in Emergent Materials



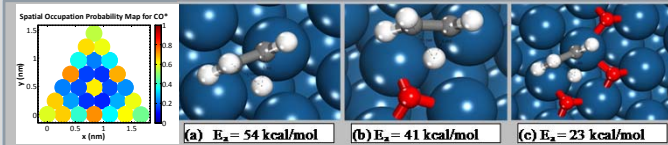
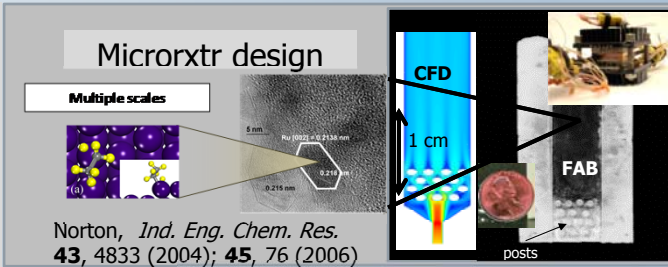
3.0 Langmuir  
NH<sub>3</sub> at 350K at  
UHV

- Ammonia decomposes on Ni-Pt:  $2\text{NH}_3 = \text{N}_2 + 3\text{H}_2$
- No decomposition on other surfaces
- Ni-Pt is the most active catalyst

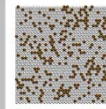
Hansgen, Chen, and Vlachos, *Nature Chem.* **2**, 484-489 (2010)

## Complexity in Catalytic Modeling

- Multiscale problem/catalyst heterogeneity
- Large mechanisms
- Complex feedstocks



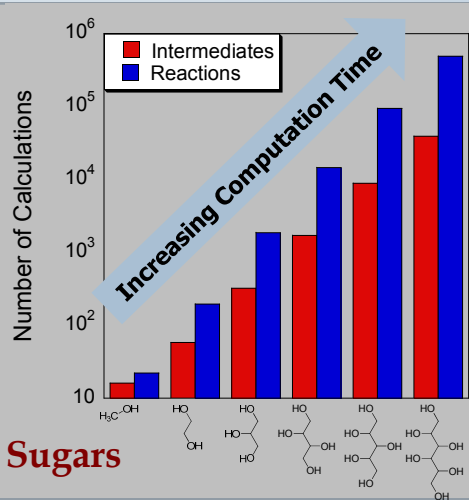
Wang et al., *J. Chem. Phys.* **133**, 224503 (2010)



Review: Saliccioli et al., *Chem. Eng. Sci.* **66**, 4319 (2011)



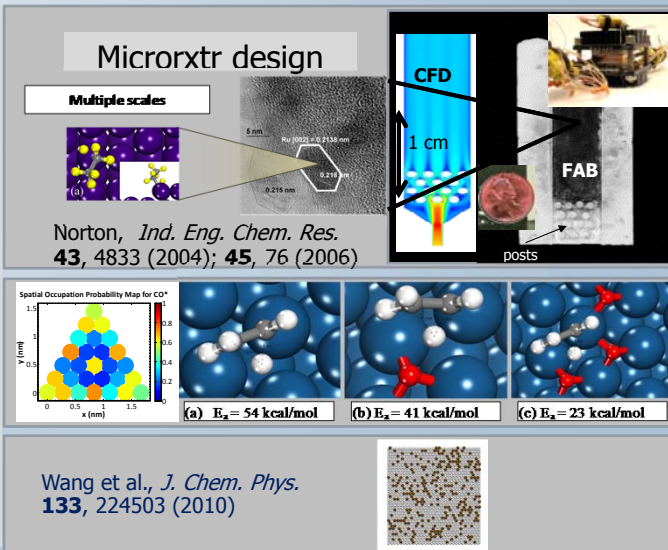
## Modeling Reactions of Large Molecules is Challenging



- Combinatorial explosion in number of calculations
- Semi-empirical methods can potentially identify relevant species and reactions instantaneously
- There is little understanding of the errors associated with these techniques**

## Complexity in Catalytic Modeling

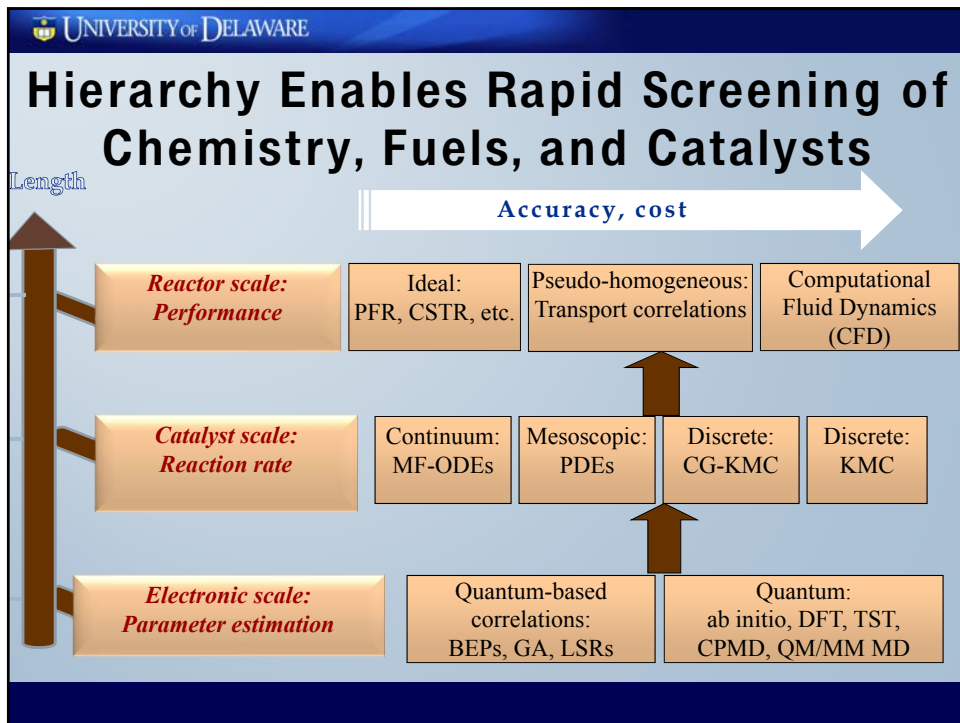
- Multiscale problem/catalyst heterogeneity
- Large mechanisms
- Complex feedstocks
- Adsorbate heterogeneity
- Combinatorial problem due to coverage effects
- Catalyst dynamics



Review: Saliccioli et al., *Chem. Eng. Sci.* **66**, 4319 (2011)

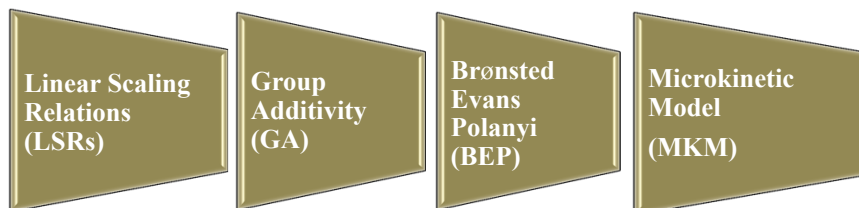
## Overarching Concept: Hierarchical Multiscale Modeling

- Start with a sufficiently simple but physically relevant model at each scale
- Link all models
- Perform a sensitivity analysis
- Identify important scale and parameter(s)
- Use higher level theory for this scale and parameter(s)
- Iterate

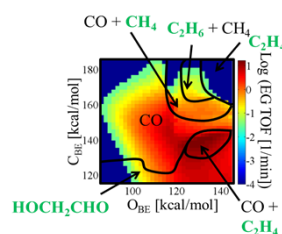




## Toward High Throughput Computing: Metal and Metal-like Catalysis



- ▶ Thermochemistry via group additivity & LSRs
- ▶ Reaction barriers and pre-exps via BEPs and TS
- ▶ Perform MKM
  - DFT-based or semi-empirical based: Screen and refine via DFT
- ▶ Assess model predictions via experiments

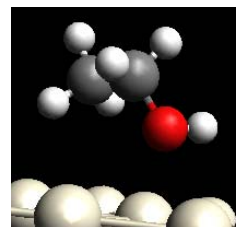


Saliccioli, Y. Chen, and Vlachos, *J. Phys. Chem. C* **114**, 20155 (2010); Saliccioli et al., *J. Phys. Chem. C* **116**, 1873 (2012)

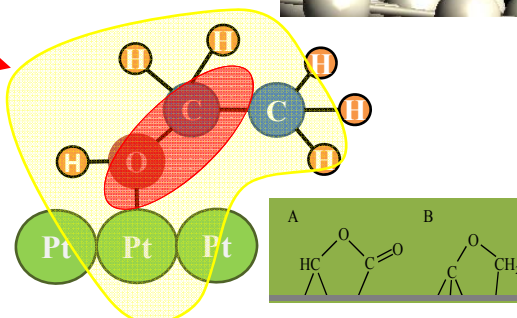


## Group Additivity for Adsorbed Species

- Binding for oxygenates traced to (CH<sub>y</sub>O<sub>x</sub>) groups
- Use alcohols and dehydrogenated intermediates of alcohols, esters, ethers, acids to develop groups
- Include contributions for  $\Delta H_{f,298}$ ,  $S(T)$  and  $C_p(T)$

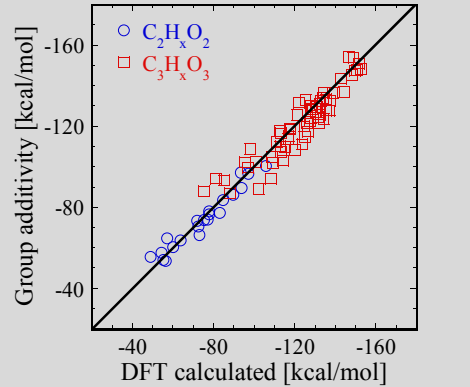


Group	$\Delta H_{f,298}$ Value [kcal/mol]
[C(C,H <sub>2</sub> )-O(M,H)]	-50.2
[C(C,H,M)-O(H)]	-46.8
[C(C,M <sub>2</sub> )-O(H)]	-39.3
[C(C,H <sub>2</sub> )-O(M)]	-26.1
[C(C,H,M)-O(M)]	-26.5
[C(C,M,M)-O(M)]	-33.4
[C(C,M)=O]	-33.4
[C(C <sub>2</sub> ,H)-O(M,H)]	-51.1
[C(C <sub>2</sub> ,M)-O(H)]	-42.9
[C(C <sub>2</sub> ,H)-O(M)]	-25.1
[C(C <sub>2</sub> ,M)-O(M)]	-23.0



## Group Additivity for Adsorbed Species

- 2<sup>nd</sup>-order effects included:
  - 4-member ring strain
  - Weak interactions
  - Hydrogen bonding
- Calculations of  $\Delta H_{f,298}$  of  $C_2H_xO_2$  and  $C_3H_xO_3$  species show good quantitative agreement

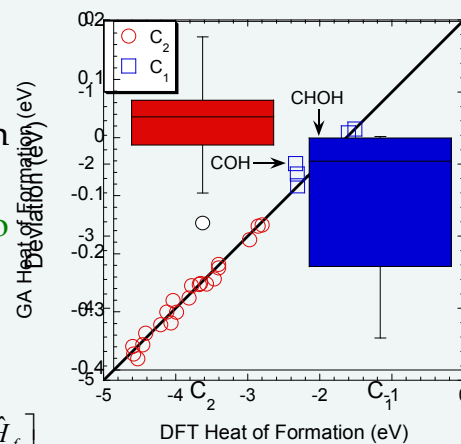


This method can be used for initial screening of larger hydrocarbons and different oxygenates (alcohols, polyols, esters, ethers, acids) found in bio-oil

Salciccioli et al., *J. Phys. Chem. C* **114**, 20155 (2010); *J. Phys. Chem. C* **116**, 1873 (2012)

## Estimated Heats of Formation May Show Systematic Deviations

- Regressed groups show errors which are normally distributed with zero mean
- **Unregressed  $C_1$  oxygenate group values have nonzero mean**



Species Error Type	$\epsilon_{H_f}$ Mean	$\hat{H}_f$ Std. Dev.	No. Points
$C_2$	0.02	0.08	21
$C_1$	-0.11	0.13	

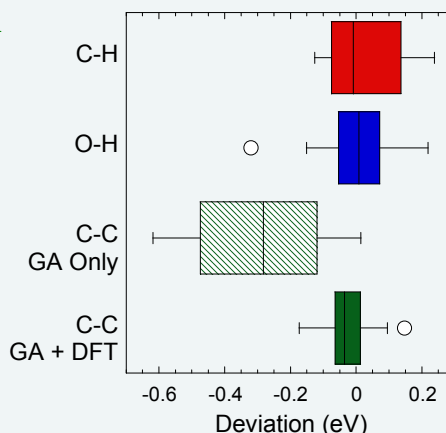
$$Var[\epsilon_{H_f}] = Var[H_f] + Var[\hat{H}_f] - 2Cov[H_f, \hat{H}_f]$$

Sutton and Vlachos, *J. Cat.*, Submitted

## Estimated Heats of Reaction Inherit Systematic Errors in Parent Species

- Bias in heats of formation propagates to heats of dissociation reactions forming  $C_1$  species
- DFT values for  $C_1$  species eliminate this bias

Reaction	Mean	Std. Dev.	No. Points
C-H	0.01	0.10	24
O-H	0.01	0.13	18
C-C GA Only	-0.28	0.20	21
C-C GA + DFT	-0.02	0.08	21



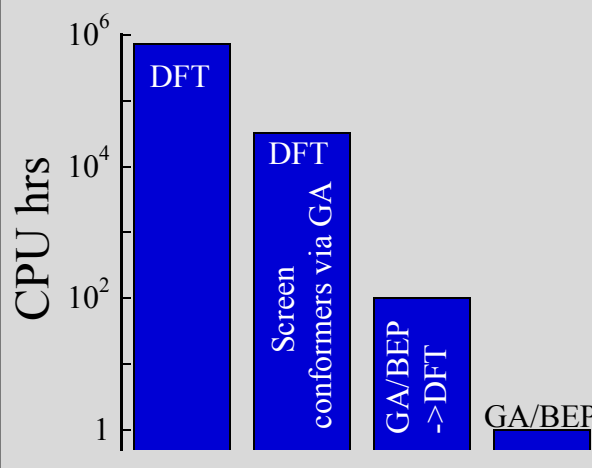
Sutton and Vlachos, *J. Cat.*, Submitted



## Computational Savings from Hierarchy

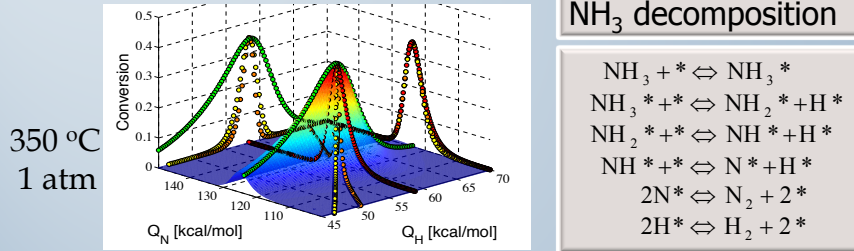
- Profound computational savings
- **(Important)** information content remains the same

### Glycerol thermal decomposition



Chen et al., *J. Phys. Chem.* **115(38)**, 18707 (2011)

## High Throughput Multiscale Model-based Catalyst Design

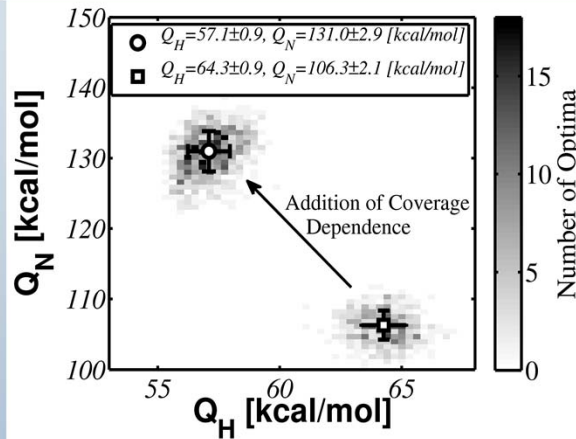


- Search is done on atomic descriptors while running the full chemistry and reactor models
- Optimal catalyst properties are identified

Prasad et al., *Chem. Eng. Sci.* **65**, 240 (2010)

## Parametric and Model-Hierarchy Uncertainty

### Effect of adsorbate-adsorbate interactions



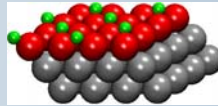
Ulissi et al., *J. Catal.* **281**, 339 (2011)

# Identifying Bimetallic Catalysts

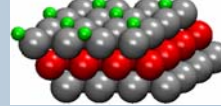


Metals	BE <sub>N</sub> (kcal/mol)
PtTiPt	56.5
PtVPt	59.5
PtCrPt	72.6
PtMnPt	84.9
PtFePt	83.9
PtCoPt	87.0
PtNiPt	89.8
<b>NiPtPt</b>	<b>137.5</b>
CoPtPt	159.9
FePtPt	169.9
MnPtPt	162.2
CrPtPt	166.5
VPtPt	184.1
TiPtPt	191.5

Pt	102.1
Ni	113.8

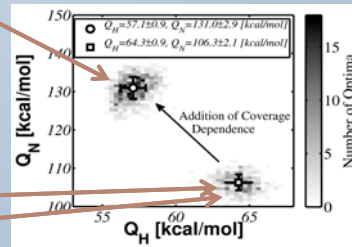


Surface

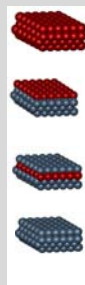
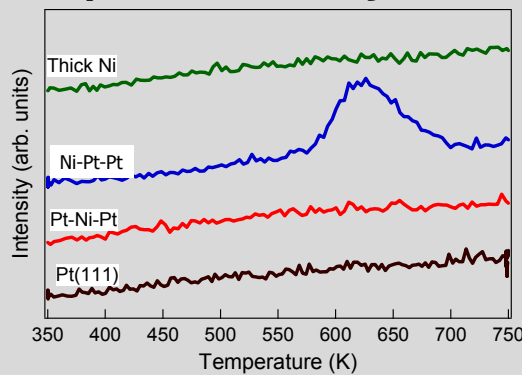


Subsurface

- Optimum heat of chemisorption of N of ~130 kcal/mol
- NiPtPt is a good prospective bimetallic surface



# Emergent Behavior Verified Experimentally for the First Time



3.0 Langmuir  
NH<sub>3</sub> at 350K at  
UHV

- Ammonia decomposes on Ni-Pt:  $2\text{NH}_3 = \text{N}_2 + 3\text{H}_2$
- No decomposition on other surfaces
- **Ni-Pt is the most active known catalyst**

Hansgen, Chen, and Vlachos, *Nature Chem.* **2**, 484-489 (2010)

## Open Questions

- Uncertainty quantification
  - What data and at which scale to inject data?\*
  - Beyond continuum models: e.g., Stochasticity
- Molecular structure and composition of catalyst
  - Optimize catalyst loading to explore bifunctionality
  - Extend to multicomponent catalysts
  - Account for crystal structure beyond core-shell
- Consider catalyst stability and catalyst dynamics
- Construct and search databases
  - Structure, composition, chemistry, reactor operating conditions

\* Prasad et al., *Ind. Eng. Chem. Res.* **47**, 6555 (2008); **48**, 5255 (2009)

## Outline

- An overview of multiscale modeling
- Drivers of nanomaterials research/New frontiers
- Control in self-assembly
- Design of catalytic materials for energy and sustainability
- **Outlook**





## Conclusions/Outlook

- ▶ Multiscale modeling of prototype models is mature
- ▶ Inverse engineering (optimization and control) at the nanoscale is challenging and offers opportunities
- ▶ Many interesting problems are either complex or emergent and their modeling is plagued by combinatorial complexity
- ▶ Hierarchical multiscale modeling is an approach coping with these issues
- ▶ Error quantification, code verification and validation, and uncertainty propagation are essential
- ▶ Parallelization holds promise for substantial speedup

## Acknowledgements

- ▶ Students/Postdocs
  - Materials design: Abhijit Chatterjee; Jake McGill; Nasser Abukhdeir
  - Energy/Catalysis: Danielle Hansgen; Ying Chen; Mike Salciccioli; Vinay Prasad
- ▶ Collaborators
  - Markos Katsoulakis (UMass) and Petr Plechac (UD); Mathematical underpinnings
  - Michela Taufer (UD); GPUs
  - Tunde Ogunnaike (UD); Control
- ▶ Funding
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